CLAIMS

***	•			
1 \ \ \ \ \	\sim	201	m	٠
We	· •	ıaı	111	٠

1 1. (Currently amended) A polymer electrolyte comprising: a modified polymeric material, said modified polymeric material including a 2 halogen containing polymer having an enhanced halogen level, said enhanced halogen level 3 relative to a halogen content of an unmodified said halogen containing polymer formed from 4 5 polymerization of its monomer; 6 a salt of an alkali metal; and an aprotic solvent, wherein said salt and said aprotic solvent are integrated with 7 said modified polymeric material as a homogeneous material. 8 (Currently amended) The polymer electrolyte of claim 1, wherein said unmodified 2. 1 halogen containing polymer includes at least one chlorine containing polymer. 2 1 3. (Original) The polymer electrolyte of claim 2, wherein said chlorine containing 2 polymer is polyvinylchloride (PVC). (Currently amended) The polymer electrolyte of claim [[3]] 1, wherein said 1 4. 2 polyvinylchloride (PVC) is suspension polyvinylchloride (PVC) a lithium ion conductivity of said polymer electrolyte at 25 C is between 0.01 S/cm² and .108 S/cm². 3

- 1 5. (Currently amended) The polymer electrolyte of claim [[3]] 1, wherein said
 2 polyvinylchloride (PVC) is emulsion polyvinylchloride (PVC) a lithium ion conductivity of said
 - 3 polymer electrolyte at 25 C is between 0.066 S/cm² and .108 S/cm².
 - 1 6. (Original) The polymer electrolyte of claim 1, wherein said modified polymeric
- 2 material comprises C-PVC, said C-PVC having 60-72 wt % chlorine.
- 7. (Original) The polymer electrolyte of claim 6, wherein said polymer electrolyte
- 2 comprises 10-40 wt % of said C-PVC.
- 1 8. (Original) The polymer electrolyte of claim 1, wherein said alkali metal salt is at
- 2 least one selected from the group consisting of LiClO₄, LiBF₄, LiAsF₆, LiPF₆, LiCF₃SO₃ and
- 3 $LiN(CF_3SO_2)_2$.
- 9. (Original) The polymer electrolyte of claim 1, wherein said electrolyte comprises
- 2 from 3-20 wt % of said salt of an alkali metal.
- 1 10. (Original) The polymer electrolyte of claim 1, wherein as said aprotic solvent is at
- 2 least one selected from the group consisting of propylene carbonate, ethylene carbonate,
- 3 dimethyl carbonate, gamma-butyrolactone, 1,3-dioxolane and dimethoxyethane.

(Original) The polymer electrolyte of claim 1, wherein said electrolyte comprises 11. 1 40-82 wt % of said aprotic solvent. 2 12. (Currently amended) A rechargeable battery, comprising: 1 an anode containing an alkali metal; 2 a cathode; and 3 a polymer electrolyte formed from a modified polymeric material, said modified 4 polymeric material including a halogen containing polymer having an enhanced halogen level, 5 said enhanced halogen level relative to a halogen content of an unmodified said halogen 6 containing polymer formed from polymerization of its monomer, a salt of an alkali metal and an 7 aprotic solvent, wherein said salt and said aprotic solvent are integrated with said modified 8 -9 polymeric material as a homogeneous material. (Currently amended) The rechargeable battery of claim 12, wherein said 1 13. unmodified halogen containing polymer comprises at least one chlorine containing polymer. 2 (Original) The rechargeable battery of claim 13, wherein said modified polymeric 14. 1 2 material comprises chlorinated polyvinylchloride (C-PVC). (Original) The rechargeable battery of claim 12, wherein in said anode comprises 1 15.

1.3

2

lithium.

(Currently amended) The rechargeable battery of claim 12, wherein said anode 16. 1 comprises a lithium alloy a lithium ion conductivity of said polymer electrolyte at 25 C is 2 between 0.01 S/cm² and .108 S/cm². 3 17. (Currently amended) The rechargeable battery of claim 16, wherein a lithium ion 1 conductivity of said polymer electrolyte at 25 C is between 0.066 S/cm² and .108 S/cm² as said 2 lithium alloy is at least one selected from the group consisting of lithium-aluminum, lithium-3 aluminum-silicon, lithium-aluminum-cadmium, lithium-aluminum-bismuth and lithium-4 5 aluminum-tin. (Currently amended) The rechargeable battery of claim 12, wherein said anode 18. 1 ggill . 2 comprises a lithium-ion intercalation material. (Original) The rechargeable battery of claim 12, wherein said cathode comprises a 19. 1 2 metal oxide. 20. (Original) The rechargeable battery of claim 12, wherein said cathode comprises a 1 2 lithium-transition metal oxide. (Original) The rechargeable cell of claim 12, wherein said cathode is at least one 21. 1

selected from the group consisting of MnO₂, LiMn₂O₄ and vanadium oxides (V_xO_y).

2

(Original) The rechargeable cell of claim 12, wherein said cathode comprises a 22. 1 2 organic polymer. (Original) The rechargeable cell of claim 12, wherein said cathode is at least one 23. 1 selected from the group consisting of polyviologen, polyacetylene and polypyrrole. 2 (Original) The rechargeable cell of claim 12, wherein said cathode comprises a 1 24. 2 sulfur containing material. (Original) The rechargeable cell of claim 12, wherein said cathode is at least one 1 25. 2 selected from the group consisting of TiS₂, S, polysulphide and polythiophene.

(Cancelled)

26.

1

A. 5 -

1 27. (Cancelled)

1 28. (Cancelled)

1 29. (Cancelled)

1 30. (Cancelled)

1 31. (Cancelled)

1 32. (Cancelled)

1 33. (Cancelled)

1 34. (Cancelled)

- 35. (Cancelled)
- 36. (Cancelled)